FLAVONOID GLYCOSIDES AND SULPHATES FROM THE DILLENIACEAE

ALBERTO A. GURNI*‡, WILFRIED A. KÖNIG† and KLAUS KUBITZKI*

* Institut für Allgemeine Botanik und Botanischer Garten, Universität Hamburg; † Institut für Organische Chemie und Biochemie der Universität Hamburg, 2000 Hamburg, West Germany

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Abstract—In a leaf survey of sixty species from eight genera of the Dilleniaceae, the following flavonoids were characterized: myricetin 3,7,3',4'-tetramethyl ether, mearnsetin 3-rhamnoside, ombuin 3,3'-disulphate, isorhamnetin 3,7,4'-trisulphate, kaempferol 3,7,4'-trisulphate and apigenin 7-galactosidesulphate.

INTRODUCTION

The Dilleniaceae are a family of tropical trees, shrubs, climbers and herbs. Interest in its taxonomy led to an earlier survey of many species for leaf flavonoid aglycones [1]. Flavonoids previously characterized in the family include rhamnocitrin, the 5-methyl ethers of quercetin and kaempferol [2], and kaempferide [1,3]. The latter substance has since been shown to be present in vivo as the corresponding dihydroflavonol [4]. The two subfamilies Tetraceroideae and Dillenioideae are characterized by the frequent occurrence of flavonoids with methylation at the 7 and 4' positions [1]. However, in a recent taxonomic treatment of the Dilleniaceae genera (K. Kubitzki, unpublished results), it was shown that the distribution of characters of taxonomic relevance within the family is so reticulate that it does not permit the definition of taxa at tribal or subfamily level. This situation led to a re-investigation of the flavonoids of the Dilleniaceae, extending the analysis to include glycosides which previously had not been studied [5].

RESULTS AND DISCUSSION

A total of 248 samples covering 60 species and 8 genera was investigated. Especially well represented are the neotropical genera *Davilla* and *Doliocarpus* and the pantropical genus *Tetracera* of which a large proportion of species were studied, together with the monotypic New World genera *Curatella* and *Pinzona*. Three species of *Schumacheria*, some *Dillenia* species and one *Acrotrema* species, all from Ceylon, were also included. Material of the predominantly Australian genus *Hibbertia* was available but was not studied because of the absence of a modern and comprehensive taxonomic framework such as is available for the rest of the family.

Here we report only the isolation of some new and interesting substances while a systematic evaluation of the

data will be given elsewhere [6]. The compounds include flavonoid glycosides, glucuronides and sulphates and one *O*-methylated aglycone.

Myricetin 3,7,3',4'-tetramethyl ether, previously identified in Cistus monspeliensis [7], was found in Doliocarpus amazonicus subsp. duckeanus, a woody climber of central Amazonia. Although myricetin occurs frequently in the Dilleniaceae, we were unable to detect the tetramethyl ether in any other species of the family. Another unusual flavonol, mearnsetin 3-rhamnoside, hitherto known from Acacia mearnsii [8] and Dorycnium suffruticosum [9] was identified in only one species, Doliocarpus spraguei, where it co-occurs with quercetin and myricetin 3-rhamnosides. Rhamnocitrin 3-glucuronide was characterized in the two African species Tetracera rosiflora and T. rutenbergii. It has previously only been recorded from Verbesina myricephala [10].

The occurrence in plants of common flavones and flavonols in a conjugated form, covalently linked to inorganic sulphate, is a relatively recent discovery [11] but now well documented [12]. Up to the present, flavonoid sulphates were unknown from the Dilleniaceae, but in the course of this study eleven sulphates have been found in the family. These include the 3-sulphates of quercetin and kaempferol, the 7-sulphates of apigenin and luteolin, myricetin 3-rhamnoside sulphate, ombuin disulphate, isorhamnetin 3,7,4'-trisulphate, kaempferol 3,7-disulphate, kaempferol 3,7,4'-trisulphate, rhamnocitin 3-sulphate and apigenin 7-galactosidesulphate. Details of new or rare compounds are given in Tables 1 and 2. Myricetin rhamnosidesulphate has been found previously in Davidsonia pruriens [13], while kaempferol 3,7-disulphate is known from Reaumuria mucronata [14] and rhamnocitrin 3-sulphate from Ammi visnaga [15]. Ombuin disulphate, isorhamnetin 3,7,4'-trisulphate, kaempferol 3,7,4'-trisulphate and apigenin 7-galactosidesulphate are new substances.

A close association between O-methylation and sulphate formation has been noted previously [11], and it is therefore not surprising to find among the flavonoid sulphates of the Dilleniaceae three derivatives of O-methylated aglycones. Moreover, when considering the occurrence of ellagic acid, flavonoid methyl ethers and

[†] Present address: University of Buenos Aires, Facultad de Farmacia y Biochímica, Cátedra de Botánica y Museo de Botánica "Juan A. Domingues".

Table 1. Chromatographic and electrophoretic values and botanical source of flavonoid sulphates

	Flavonoid sulphate	nnosidesulphate	ate	',4'-trisulphate	Jisulphate			'-trisulphate	sulphate				tosidesulphate	
	Fla	Myricetin 3-rhamnosidesulphate	Ombuin disulphate	Isorhamnetin 3,7,4'-trisulphate	Kaempferol 3,7-disulphate			Kaempferol 3,7,4'-trisulphate	Rhamnocitrin 3-sulphate				Apigenin 7-galactosidesulphate	
	Botanical source	Davilla macrocarpa, D. flexuosa	Acrotrema uniflorum	Acrotrema uniflorum	Schumacheria	castaneifolia,	Dillenia bracteata	Acrotrema uniflorum	Tetracera poggei,	T. rosiflora,	T. rutenbergii,	T. alnifolia	Tetracera	stuhlmanniana
	Mobility*	99'0	16.1	2.34	2.17			2.38	1.26				Not	measured
_	О ² Н	88	78	8	94			94	82				37	
R_f (\times 100) in	HOAc (15%)	62	20	80	72			8	8				23	
R	BAW	31	37	7	11			7	42				33	
	Compound No.	1	2	3	4			S	9				7	

^{*} Mobility relative to quercetin 3-sulphate, run on paper Schleicher & Schüll No. 2316 in formate-acetate buffer, pH 2.2 at 40 V/cm for 1 hr.

Table 2. UV spectra of flavonoid sulphates

banoamo	Me	НО	NaOMe	Ме	NaOAc	Ac	NaOAc/H ₃ BO ₃	/H ₃ BO ₃	AIC	AICI ₃	AICI	AIC13/HCI
No.*	Band II	Band I	Band [] Band I	Band I	Band II Band I	Band I	Band II Band I	Band I	Band II Band I	Band I	Band II Band I	Band I
-	263	365	жр	dec	270	400	270	368	274	426	278	009
2	255	353	766	398	259	356	260	356	268	900	396	9
3	254	350	268	376	260	355	254	352	275	400	274	36%
4	566	350	276	394	270	352	270	353	277	356	277	350
S	250 270	344	280	394	270	344	270	34.	278	600	278	400
9	506	365	275	380	274	375	270	356	275	400	275	60
7	566	330	270	384	799	390	265	336	276	380	275	380

^{*} Numbering of substances as in Table 1.

flavonoid sulphates there are some similarities between the Dilleniaceae, Cistaceae, Guttiferae, Bixaceae and other families [16] which may be of some systematic significance.

EXPERIMENTAL

Plant material. This study is based mainly on air-dried material collected by the third author in South America and Ceylon. All determinations were checked and vouchers were deposited at HBG and M. The study was supplemented by the investigation of herbarium material obtained from the herbaria at BR, M, NY, U, US. A complete list of the plant material studied is available on request from the third author. In six cases, extracts from leaves preserved in EtOH were compared with those from air-dried leaves but no difference was noticed.

Flavonoid identifications. Plant material was extracted with boiling H₂O, evapd under red. pres., dissolved in MeOH and the extracts were analysed using standard procedures.

Myricetin 3,7,3',4'-tetramethyl ether has $\lambda_{m}^{\text{MeOH}}$ nm: 265, 343; +NaOMe 270, 370 (stable): +NaOAc 267, 344; +AlCl₃ 277, 398; R_f in BAW 0.93; in CHCl₃-HOAc (2:1, H₂O-saturated) 0.99 and in Forestal 0.90. MS showed a molecular ion m/z 374 corresponding to a tetramethylated myricetin derivative. UV data showed that only the 5- and 5'-hydroxyls were free.

Mearnsetin 3-rhamnoside has $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 264, 352; +NaOMe 280, 398 (stable): +NaOAc 276, 380: R_f in BAW 0.90, in HOAc 15% 0.68, in H₂O 0.28. Acid hydrolysis yielded mearnsetin and rhamnose. Mearnsetin had $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 264, 366; +NaOMe 280, 414 (stable); +NaOAc 276, 380; R_f in BAW 0.74, in CHCl₃-HOAc 0.23 and in Forestal 0.40. This chromatographic behaviour is indicative of a derivative of myricetin which was confirmed by MS, m/z (rel. int.): M 332 (100), M - Me 317 (97), M - Me - CO 289 (25), ion B 167 (4), ion A 153 (11). UV shows that only the 4' hydroxyl is blocked.

Rhamnocitrin 3-glucuronide has $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 266, 352; +NaOMe 270, 394 (stable); +NaOAc 267, 363. R_f in BAW 0.37, in HOAc 15% 0.55 and in H₂O 0.80. Hydrolysis yielded rhamnocitrin and glucuronic acid. Rhamnocitrin had $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 266, 366; +NaOMe 267, 425 (decomposing); +NaOAc 266, 385. R_f in BAW 0.74, in CHCl₃-HOAc 0.73 and in Forestal 0.53. MS of the aglycone, m/z (rel. int.): M 300 (100), M - H 299 (54), M - Me 285 (2), M - Me - CO 257 (38), ion A 167 (9), ion B 121 (66) confirmed the presence of kaempferol 7-O-methyl ether. UV of the glucuronide indicated an additional blocking of the hydroxyl in the 3-position.

Flavonoid sulphates. These compounds showed characteristic arrow-shaped spots on the chromatograms, and hydrolysis gave aglycone, sulphate and potassium ions plus a sugar in the case of sulphated glycosides. The presence of sulphate was confirmed by precipitation with BaCl₂. Known sulphates were identified by acid hydrolysis to aglycone and sulphate, UV analysis and by comparison with an authentic sample when available.

Myricetin 3-rhamnosidesulphate. UV showed that only the 3-hydroxyl was blocked. Hydrolysis yielded myricetin, rhamnose and sulphate. Since only the 3-hydroxyl is blocked, the sulphate must be linked to the sugar moiety.

Ombuin disulphate on acid hydrolysis gave ombuin, which had $\lambda_{\text{mas}}^{\text{MeOH}}$ nm: 252, 370; + NaOMe 252, 432 (stable); + NaOAe 252, 354. R_f in BAW 0.94, in CHCl₃-HOAe 0.98 and in Forestal 0.87. MS of the aglycone, m/z (rel. int.): M 330 (100), M - H 329 (16), M - Me 315 (6). M - Me - CO 287 (13), ion A 167 (7), ion B 151 (12). In comparison with the aglycone, UV of the sulphate indicated that the 3-position was blocked and since the electrophoretic mobility of this substance is higher than that of a monosulphate, a second sulphate is probably present attached either to the first or at the 3'-position.

Isorhamnetin 3,7,4'-trisulphate gave isorhamnetin on acid hydrolysis. The UV spectrum of the sulphate showed blocking of hydroxyls in the 3-, 7- and 4'-positions.

Kaempferol 3,7,4'-trisulphate also yielded kaempferol and sulphate ions on acid hydrolysis but the UV spectrum showed blocking of hydroxyls in the 3-, 7- and 4'- positions.

Apigenin 7-galactosidesulphate. After hydrolysis, apigenin, galactose and sulphate ions were detected. UV showed that only the 7-hydroxyl was blocked indicating that the sulphate is attached to the sugar moiety at this position.

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